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Diels-Alder Reactions of (Z)-Ethyl 3-[(1-ethoxycarbonyloxy-2-methoxy)ethenyl]-2-(ethoxycarbonyloxy)indole-1-carboxylate. Synthesis of the Carbazole Alkaloid Carbazomycin B.

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Abstract: Diels-Alder reactions of the title 3-vinylindole 2 with N-phenylmaleimide, maleimide, DEAD and DMAD are described. From Compound 10, obtained from 2 and DMAD, Carbazomycin B (1) was prepared.

We have recently reported 1 a new synthesis of 3-vinylindoles starting from indol-2(3H)one. We now report the use of (Z)-ethyl 3-[(1-ethoxycarbonyloxy-2-methoxy)ethenyl]-2-(ethoxycarbonyloxy)indole-1-carboxylate (2) 1 as a diene in the Diels-Alder synthesis of Carbazomycin B, following the strategy by Pindur 2 for the synthesis of 4-demethoxy carbazomycin.

Carbazomycin B $(1)^3$ is an inhibitor of 5-lipoxygenase⁴ and posseses weak antibacterial and antiyeast activity.^{3a} It also inhibits the growth of some phytopathogenic fungi^{3a} and several syntheses of the alkaloid system have been described.⁵

Compound 2 reacts in boiling toluene with N-phenylmaleimide 3a (NPMI) to furnish the carbazole 4a (Scheme 1).

Scheme 1

OMe

Scheme 1

OR1

$$OR1$$
 $OR1$
 $OR1$

Reaction with diethyl azodicarboxylate (DEAD) at room temperature gave compound 5 which on alkaline hydrolysis produces the corresponding pyridazinoindole 6. (Scheme 2).

Scheme 2

Compound 5 probably arises from the Diels-Alder adduct *via* the hydrolysis of the enolcarbonate and successive CO₂ and EtOH elimination as shown in Scheme 2. This hypothesis is strongly supported by results we have obtained with Diels-Alder adducts from the diene 2 and maleimide 3a,b under sonication conditions. Indeed, if the reaction between the diene 2 and NPMI 3a is carried out in an ultrasound bath, the *endo* adduct 7a is obtained in very high yield (Scheme 3). The adduct 7a is easily obtained in a pure state by crystallization from Et₂O of the reaction mixture.

By treatment with triethylamine in CH₂Cl₂ solution at r.t. 7a is quantitatively transformed into the carbazole 4a (Scheme 3).

Scheme 3

When a CH₂Cl₂ solution of the compound **7a** is treated at 0 °C with a catalytic amount of H₂SO₄, the new derivative **8a** is obtained. The analysis of the reaction mixture shows the presence of an intermediate which is thermally transformed into compound **8a** with elimination of CO₂ and EtOH. A possible hypothesis on the structure of this intermediate is shown in Scheme 3.

Compound **7b**, from Diels-Alder reaction with the maleimide **3b** was not isolated in pure state. The crude reaction mixture was submitted to acidic treatment to give directly **8b** in 70% yield, based on the diene **2**.

The structure of all new compounds is assigned on the basis of analytical and spectroscopic data, and in the case of **8b**, also by diffraction analysis⁶. Figure 1 shows an ORTEP view of the molecule with the atomic numbering scheme of the heavy atoms.

When compounds **8a,b** are treated with bromine in dichloromethane solution at r.t., bromination followed by HBr elimination gives the carbazole **9a,b** (Scheme 3) in very good yields. Any attempts to open, in acceptable yields, the phtalimide ring were unsuccessfully.

Figure 1. ORTEP of 8b.

Very good yields (90%) were also obtained in the Diels-Alder reaction of the diene 2 with dimethyl acetylendicarboxylate (DMAD), giving the carbazole 10. Using from 10, carbazomycin B 1 may be obtained in three steps from 2 in an overall yield of 50% (Scheme 4). The spectral data for the synthetic carbazomycin B agree with those described in the literature.^{3,5} Attempted acidic hydrolysis of the acid 11 gives raise to decarboxylation and the ester 12 is obtained in 58% yield, confirming the structure of compound 11 (Scheme 4).

Scheme 4

HO

OMe

$$20\% \text{ H}_2\text{SO}_4 \text{ aq, MeOH}$$
 Δ , 20h, 58%

COOMe

EXPERIMENTAL

Melting points were determined on a Buchi apparatus and are uncorrected. IR spectra were recorded on a Perkin-Elmer 298 instrument, in nujol mull for solids and as liquid film for oils. ¹H-NMR were recorded on a Varian Gemini 200 spectrometer in CDCl₃ solution unless otherwise stated; chemical shifts are expressed in ppm (δ) relative to TMS, coupling constants (J) in Hz. Colomn chromatography was performed on Kieselgel Merck 60, 0.063-0.2 mm. Evaporation was carried out under vacuum in a rotary evaporator.

5-Ethoxycarbonyloxy-1,3-dioxo-2-phenyl-2,3-dihydro-1H-pyrrolo[3,4-a]carbazole-10-carboxylic acid ethyl ester **4a**.

Compound **2** (1 mmol, 421 mg) and **3a** (1.2 mmol, 207 mg) were dissolved in toluene (30 mL). After heating to reflux for 48h, the mixture was evaporated and the residue purified by silica gel column chromatography (hexane-dichloromethane, 2:1) affording compound **4a** (345 mg, 73%); mp 180-181 °C (CH₂Cl₂-Et₂O); 1 H NMR $^{\circ}$ 1.36 (3H, t, 7.2), 1.48 (3H, t, 7.1), 4.46 (2H, q, 7.1), 4.55 (2H, q, 7.2), 7.39-7.54 (5H, m), 7.62 (1H, t, 8.8), 7.92 (1H, s), 8.19 (2H, d, 8.8); IR 1775, 1738, 1714 cm⁻¹; Anal. Calcd. for C₂₆H₂₀N₂O₇: C, 66.10; H, 4.27; N, 5.93. Found C, 66.28; H 4.19; N, 5.81.

1,2,9-Tri-ethoxycarbonyl-3-methoxy-4-oxo-1,2,3,4-tetrahydro-pyridazino[3,4-b]indole 5.

Compound **2** (1 mmol, 421 mg) and diethyl azodicarboxylate (DEAD) (1.2 mmol, 207 mg) were dissolved in CH₂Cl₂ (20 mL). After 48h at r.t., the mixture was purified by silica gel column chromatography (hexane-dichloromethane, 2:1) affording compound **5** (oil, 410 mg, 98%); ¹H-NMR δ 1.32 (6H, m), 1.47 (3H, t, 7.2), 3.60 (3H, s), 4.33 (4H, m), 4.51 (2H, q, 7.2), 5.54 (1H, bs), 7.35 (2H, m), 8.12 (2H, m); IR 1745, 1727, 1718 cm⁻¹; Anal. Calcd. for C₂₀H₂₃N₃O₈: C, 55.42; H, 5.35; N, 9.70. Found C, 55.30; H, 5.40; N, 9.89.

1,4-Dihydro-4-oxo-9H-pyridazino[3,4-b]indole 6.

Compound 5 (1 mmol, 433 mg) was dissolved in MeOH (20 mL) and H₂O (20 mL) and NaOH (6 mmol, 240 mg) was then added. The mixture was heated under reflux for 30 min. MeOH was evaporated and the solid filtered and crystallized gave compound 6 (120 mg, 65%); mp 227-230 °C (CH₂Cl₂-Et₂O); 1 H-NMR (DMSO-d₆) 6 6.15 (1H, bs, exchange with D₂O), 7.34 (1H, t, 6.7), 7.60 (2H, m), 8.18 (1H, d, 7.8), 8.26

(1H, s), 12.55 (1H, bs, exchange with D2O); IR 3200, 3150, 1605 cm $^{-1}$; Anal. Calcd. for C₁₀H₇N₃O: C, 64.86; H, 3.81; N, 22.69. Found: C, 64.70; H, 3.90; N, 22.70.

 $5,10a\beta$ -Bis-ethoxycarbonyloxy- 4α -methoxy-1,3-dioxo-2-phenyl- $2,3,3a\beta,4,10a,10b\beta$ -hexahydro-1H-pyrrolo[3,4-a]carbozylic acid ethyl ester 7a.

A mixture of compound **2** (5 mmol, 2.1 g) and N-phenylmaleimide **3a** (6 mmol, 1.03 g) in CH₂Cl₂ (3 mL) was sonicated in a sonication bath at 25-50 °C for 20h. After this time the evaporation of the solvent gives a solid residue that, was taken up in Et₂O (50 mL) and filtered giving pure **7a** (2.67 g, 90%); mp 179-180 °C (CH₂Cl₂-Et₂O); ¹H-NMR δ 1.23 (3H, t, 7.2), 1.38 (3H, t, 7.2), 1.46 (3H, t, 7.1), 3.39 (3H, s), 4.03 (1H, dd, 3.4, 8.3), 4.14 (2H, m), 4.34 (2H, m), 4.52 (2H, m), 5.07 (1H, d, 8.3), 5.28 (1H, d, 3.4), 7.24-7.38 (5H, m), 7.40-7.52 (2H, m), 8.09 (1H,m), 8.21 (1H, m); IR 1762, 1748, 1730, 1715 cm⁻¹; Anal. Calcd. for C₃₀H₃₀N₂O₁₁: C, 60.60; H, 5.09; N, 4.71. Found: C, 60.80; H, 4.91; N, 4.90.

The compound **7b** was prepared in a similar procedure using maleimide **3b** (6 mmol, 582 mg) as a dienophile. In this case the product **7b** was used without purification.

5-Ethoxycarbonyloxy1,3-dioxo-2-phenyl-2,3-dihydro-1H-pyrrolo[3,4-a]carbazole-10-carboxylic acid ethyl ester **4a** from **7a**.

Compound **7a** (1 mmol, 595 mg) was dissolved in CH₂Cl₂ (10 mL) and then TEA was added (0.05 mL). Evaporation of the solvent and crystallization afforded pure **4a** (448 mg, 95%).

 4α -Methoxy-1,3,5-trioxo-2-phenyl-2,3,3a β ,4,5,10b β -hexahydro-1H-pyrrolo[3,4-a]carbazole-10-carboxylic acid ethyl ester **8a** and 4α -methoxy-1,3,5-trioxo-2,3,3a β ,4,5,10b β -hexahydro-1H-pyrrolo[3,4-a]carbazole-10-carboxylic acid ethyl ester **8b**.

Compound 7 (5 mmol) was dissolved in CH₂Cl₂ (60 mL) and after cooling at 0 °C two drops H₂SO₄ conc. were added. The mixture was then neutralized with NaHCO₃ and the product purified by silica gel column chromatography (dichloromethane-ethyl ether, 30:1) affording from **7a** compound **8a** (1.94 g, 90%); mp 230-232 °C (CH₂Cl₂-Et₂O); ¹H-NMR δ 1.51 (3H, t, 7.1), 3.36 (3H, s), 3.39 (1H, dd, 4.2, 8.0), 4.14 (1H, d, 4.2), 4.59 (2H, m), 5.33 (1H, d, 8.0), 7.27 (1H, m), 7.38-7.50 (6H, m), 8.11 (1H, m), 8.26 (1H, m); IR 1749, 1729, 1715 cm⁻¹; Anal. Calcd. for C₂4H₂0N₂O₆: C, 66.66; H, 4.66; N, 6.48. Found: C, 66.50; H, 4.66; N, 6.42. From **7b** compound **8b** was obtained (1.28 g, 72% based on **2**); mp 248-250 °C (CH₂Cl₂-Et₂O); ¹H-NMR δ 1.52 (3H, t, 7.2), 3.32 (3H, s), 3.82 (1H, dd, 4.2, 8.0), 4.04 (1H, d, 4.2), 4.61 (2H, m), 5.21 (1H, d, 8.0), 7.42 (2H, m), 7.97 (1H, bs, exchange with D₂O), 8.13 (1H, m), 8.23 (1H, m); IR 3180, 3070, 1782, 1740, 1720 cm⁻¹; Anal. Calcd. for C₁8H₁6N₂O₆: C, 60.67; H, 4.53; N, 7.86. Found: C, 60.59; H, 4.55; N, 7.90.

5-Hydroxy-4-methoxy-1,3-dioxo-2-phenyl-2,3-dihydro-1H-pyrrolo[3,4-a]carbazole-10-carboxylic acid ethyl ester **9a** and 5-hydroxy-4-methoxy-1,3-dioxo-2,3-dihydro-1H-pyrrolo[3,4-a]carbazole-10-carboxylic acid ethyl ester **9b**.

Compound 8 (1 mmol) was dissolved in CH₂Cl₂ (30 mL) and then Br₂ (2 mmol, 0.11 mL) was added under stirring. After 1.5 h at r.t. the solution was evaporated and the residue purified by silica gel column chromatography (dichloromethane-ethyl ether, 40:1), affording, from 8a, compound 9a (409 mg, 95%); mp

177-178 °C (CH₂Cl₂-Et₂O); ¹H-NMR δ 1.38 (3H, t, 7.1), 4.30 (3H, s), 4.54 (2H, q, 7.1), 7.09 (1H, s, exchange with D₂O), 7.39-7.61 (7H, m), 8.14 (1H, d, 7.4), 8.31 (1H, dd, 1.5, 7.7); IR 3450, 1758, 1722, 1704 cm⁻¹; Anal. Calcd. for C₂4H₁8N₂O₆: C, 66.97; H, 4.22; N, 6.51. Found: C, 67.03; H, 4.17; N, 6.40. From **8b**, compound **9b** was obtained (315 mg, 89%); mp 313-314 °C; ¹H-NMR (DMSO-d₆) δ 1.27 (3H, t, 7.1), 4.00 (3H, s), 4.42 (2H, q, 7.1), 7.48 (1H, t, 7.5), 7.61 (1H, m), 8.02 (1H, d, 8.0), 8.31 (1H, d, 6.9), 11.10 (1H, s, exchange with D₂O), 11.42 (1H, s, exchange with D₂O); IR 3390, 3160, 1740, 1735, 1690 cm⁻¹; Anal. Calcd. for C₁8H₁4N₂O₆: C, 61.02; H, 3.98; N, 7.91. Found: C, 60.94; H, 4.01; N, 8.01.

4-Ethoxycarbonyloxy-3-methoxy-carbazole-1,2,9-tricarboxylic acid 9-ethyl ester 1,2-dimethyl ester 10. Compound 2 (7 mmol, 2.95 g) and DMAD (21 mmol, 2.58 mL) were heated in a sealed tube at 125 °C for 2d. After this time the reaction mixture was purified by silica gel column chromatography (hexane-ethyl ether, 1:1), giving pure compound 10 (2.98 g, 90%); mp 93-96 °C (hexane-Et₂O); ¹H-NMR δ 1.43 (6H, m), 3.91 (3H, s), 3.95 (3H, s), 3.97 (3H, s), 4.44 (4H, m), 7.40 (1H, t, 7.6), 7.55 (1H, m), 8.01 (1H, d, 7.7), 8.16 (1H, d, 8.3); IR 1758, 1730, 1718 cm⁻¹; Anal. Calcd. for C₂₃H₂₃NO₁₀: C, 58.35; H, 4.90; N, 2.96. Found: C, 58.26; H, 4.93; N, 2.99.

4-Hydroxy-3-methoxy-9H-carbazole-1.2-dicarboxylic acid 2-methyl ester 11.

Compound 10 (4 mmol, 1.89 g) was dissolved in MeOH (50 mL) and H₂O (50 mL). NaOH (60 mmol, 3 g) was then added and the mixture heated under reflux for 3 h. MeOH was evaporated and the solution acidified with 18% HCl. The mixture was extracted with AcOEt (3 x 30 mL). the organic layer was dried, filtered and evaporated to give, after crystallization, pure compound 11 (0.934 g, 74%); mp 255-257 °C (CH₂Cl₂-Et₂O); 1 H-NMR δ 3.83 (3H, s), 3.91 (3H, s), 7.20 (1H, m), 7.38 (2H, m), 8.27 (1H, d, 7.7), 9.11 (1H, bs, exchange with D₂O), 9.93 (1H, bs, exchange with D₂O), 12.30 (1H, bs, exchange with D₂O). IR 3200-3450br, 1710 cm⁻¹; Anal. Calcd. for C₁₆H₁₃NO₆: C, 60.95; H, 4.16; N, 4.44. Found: C, 61.10; H, 4.20; N, 4.51.

Carbazomycin B 1.

Red-Al (5 mL, 3.5 M solution) was added to a solution of compound 1 1 (1 mmol, 315 mg) in anhydrous dioxane (30 mL), under nitrogen. After 4h undr reflux, the solution was cooled, MeOH (3 mL) added and evaporated. The residue was taken up with 15% HCl (50 mL) and extracted with CH₂Cl₂ (2 x 50 mL). The organic layer was dried, filtered and evaporated. Silica gel column chromatography (hexane-dichloromethane, 1:1) of the residue afforded pure 1 (180 mg, 75%); mp 160-161 °C (pentane-Et₂O); in ref. 3a reported 158.5-160 °C, in ref. 5d reported 162-164 °C.

4-Hydroxy-3-methoxy-9H-carbazole-2-carboxylic acid methyl ester 12.

Compound 11 (2 mmol, 630 mg) was dissolved in MeOH (30 mL) and 10% H₂SO₄ (20 mL). Th solution was heated under reflux for 20h. MeOH was evaporated and the residue extracted with CH₂Cl₂ (2 x 20 mL). The organic layer was dried, filtered and evaporated and the residue purified by silica gel column chromatography (dichloromethane-ethyl ether, 1:1) gave compound 12 (314 mg, 58%); mp 142-146 °C (CH₂Cl₂-Et₂O); ¹H-NMR δ 3.96 (3H, s), 3.99 (3H, s), 6.51 (1H, s, exchange with D₂O), 7.25 (1H, m),

7.43 (2H, m), 7.56 (1H,s) 8.16 (1H, bs, exchange with D₂O), 8.33 (1H, d, 7.7); IR 3480, 3340, 1698 cm⁻¹; Anal. Calcd. for C₁5H₁₃NO₄: C, 66.41; H, 4.83; N, 5.16. Found: C, 66.43; H, 4.80; N, 5.11.

X-ray structure determination of 8b.

Single crystal of **8b** was obtained from acetone-ethyl ether: the crystals are stable in sealed glass capillary. Data were collected on a SIEMENS-P4 diffractometer using $K\alpha$ radiation ($\lambda = 0.71069 \text{ Å}$).

The structure were solved by SIR92⁷ and refined by full-matrix least squares based on I (SHELXL-93)⁸. **8b** contains ethylic ether with **8b**/solvate ratio 2:1; the ether molecule is disordered around a centre. Only H atoms of N-H and of quaternary C-H refined. The disordered ether molecule in **8b** was introduced as a planar all trans model (whitout H atoms) and then refined anisotropically with some constraints (SAME and SIMU instructions of SHELX-93). Figure 1 shows the molecule in **8b**. As can be see from the Figure H3 and H7 are syn each to the other. C₁₈H₁₆N₂O₆ . 1/2 C₄H₁₀O, $F_W = 393.40$, triclinic, space group P $\bar{1}$, a = 7.888(1), b = 9.844(2), c = 12.519(2), $\alpha = 97.99(1)$, $\beta = 94.99(1)$, $\gamma = 93.76(1)$, V = 956.0(3), Z = 2, $D_{calc} = 1.367$ g.cm⁻³, data collection: $4.5 < 20 < 50.0^{\circ}$, hkl range 0.9; -11,11; -14,14, No. independent data 3328, 2117 observed $[I > 2\sigma(I)]$. Refinement on I, $R_1 = 0.056$, wR₂ (all reflections) 0.175, goodness-of-fit = 1.034, $|\Delta \rho| \min = 0.35$, $|\Delta \sigma| \max = 0.042$.

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